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Synthesis of Highly Functionalized 1-Azadienes and Ketenimines

Issa Yavari*, Hoorieh Djahaniani, and Farough Nasiri

Department of Chemistry, University of Tarbiat Modarres, P.O. Box 14115-175, Tehran, Iran

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Summary. The 1:1 reactive intermediate generated by the addition of alkyl isocyanides to dialkyl acetylenedicarboxylates was trapped by fairly strong NH-acids such as carbazole, indole, or pyrrole to yield highly functionalized 1-azadienes and ketenimines.

Keywords. 1-Azadienes; Ketenimines; Alkyl isocyanides; Acetylenic esters; NH-Acids; Ugi reaction.

Introduction

Multicomponent reactions (MCRs), by virtue of their convergence, productivity, facile execution, and generally high yields of products, have attracted much attention from the point of combinatorial chemistry. Of pivotal importance in this area are the isocyanide based MCRs such as the versatile Ugi and *Passerini* reactions [1–4]. The addition of nucleophilic carbenes such as isocyanides to dialkyl acetylenedicarboxylates has been investigated in detail by a number of research groups [5–9]. The initially formed 1:1 zwitterionic species undergo further reactions with alkyl isocyanides and dialkyl acetylenedicarboxylates, ultimately leading to a variety of complex heterocyclic systems [10, 11]. The reaction of 1-azadienes with various chiral dienophiles, which leads to substituted pyridines, has been reported [12, 13]. We wish to report here an efficient synthetic route to 1-azadienes and stable ketenimines using alkyl isocyanides 1, dialkyl acetylenedicarboxylates 2, and a fairly strong NH-acid, such as carbazole (**3a**), indole (**3b**), or pyrrole (**3c**) (Scheme 1).

Results and Discussion

The reaction of alkyl isocyanides 1 with acetylenic esters 2 in the presence of an NH-acid such as carbazole or indole afforded the isomeric 1-azadienes 4 and

^{*} Corresponding author. E-mail: isayavar@yahoo.com



highly functionalized ketenimines **5** in fairly good yields. However, using pyrrole as the NH-acid, only the 1-azadiene derivative **4** was obtained (Scheme 1).

The structures of 4 and 5 were deduced from their elemental analyses, mass spectrometric data, and their ¹H and ¹³C NMR and IR spectra. The ¹H NMR spectrum of 4a exhibited four sharp lines for *tert*-butyl ($\delta = 1.17$ ppm), metoxy $(\delta = 3.68 \text{ and } 3.70 \text{ ppm})$, and vinyl $(\delta = 6.44 \text{ ppm})$ protons. The carbazole moiety appeared as a multiplet at $\delta = 7.2 - 8.1$ ppm. The ¹³C NMR spectrum of **4a** showed signals for *tert*-butyl ($\delta = 29.68$ ppm), methoxy ($\delta = 52.10$ and 52.56 ppm), and olefinic ($\delta = 123.11$ ppm) carbons in agreement with the proposed structure. Partial assignments of these resonances are given in the Experimental. The ¹H NMR spectra of **4b**–**4h** are similar to that of **4a**, except for the signals of the cyclohexyl, ester, and heterocyclic moieties. The ¹H NMR spectrum of **5a** displayed sharp signals for the *tert*-butyl ($\delta = 1.10$ ppm), methoxy ($\delta = 3.57$ and 3.98 ppm), and methine ($\delta = 5.07$ ppm) protons. The ¹³C NMR spectrum of **5a** exhibited fifteen sharp lines in agreement with the proposed structure. Partial assignment of these resonances is given in the Experimental. The structural assignments of 5a-5e made on the basis of their NMR spectra were supported by their IR spectra. Of special interest is the strong ketenimine absorption band at about $\bar{\nu} = 2050 \,\mathrm{cm}^{-1}$.

On the basis of the well established chemistry of isocyanides [14-16] it is reasonable to assume that **4** and **5** result from initial addition of the alkyl isocyanide to the acetylenic ester and subsequent protonation of the 1:1 adduct by the NH-acid. Then, the positively charged ion can be attacked at two positions by the nitrogen atom of the anion of the NH-acid. Direct addition leads to 1-azadienes **4** and conjugate addition produces ketenimines **5** (Scheme 2).



Scheme 2

Synthesis of Highly Functionalized 1-Azadienes and Ketenimines

In conclusion, we have found a simple and efficient method for the preparation of highly functionalized ketenimines and 1-azadienes. The present method carries the advantage that not only the reaction is performed under neutral conditions, but also the starting materials and reagents can be mixed without any activation or modification.

Experimental

Melting points were measured on an Electrothermal 9100 apparatus. Elemental analyses (C, H, N) were performed using a Heraeus CHN-O-Rapid analyzer. Their results agreed favorably with the calculated values. IR spectra were measured on a Shimadzu IR-460 spectrometer. ¹H and ¹³C NMR spectra were measured with a Bruker DRX-500 AVANCE instrument with CDCl₃ as solvent at 500.1 and 125.7 MHz. Mass spectra were recorded on a Finnigan-Matt 8430 mass spectrometer operating at an ionization potential of 70 eV. Dialkyl acetylenedicarboxylates, alkyl isocyanides, carbazole, pyrrole, and indole were obtained from Fluka (Buchs, Switzerland) and were used without further purification.

General Procedure (examplified by 4a and 5a)

To a magnetically stirred solution of 0.33 g of carbazole (2 mmol) and 0.28 g of dimethyl acetylenedicarboxylate (2 mmol) in 10 cm^3 of CH₂Cl₂ 0.45 g of *tert*-butyl isocyanide (2 mmol) were added dropwise at -10° C over 10 min. The reaction mixture was then allowed to warm up to room temperature and stand for 24 h. The solvent was removed under reduced pressure and the mixture of products was obtained as a yellow oil. The two isomeric products were separated by silica gel column chromatography (Merck 230–400 mesh) using *n*-hexane:ethyl acetate as eluent. The first compound was eluted using a 9:1 mixture and identified as **4a**, subsequent elution with a 7:1 mixture gave **5a**.

Dimethyl 2-(tert-butylimino(carbazol-9-yl)methyl)but-2-enedioate (4a, C₂₃H₂₄N₂O₄)

White crystals (from diethyl ether), yield 0.45 g (58%), mp 97–98°C; IR (KBr): $\bar{\nu} = 1715$ and 1739 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.17$ (s, CMe₃), 3.68 and 3.70 (2s, 2OCH₃), 6.44 (s, CH), 7.27–8.12 (m, 8CH) ppm; ¹³C NMR: $\delta = 29.68$ (CMe₃), 52.10 and 52.56 (2OCH₃), 59.07 (CMe₃), 110.47, 120.23, 120.68, and 126.69 (8CH), 122.39 and 140.11 (4C), 123.11 (CH vinyl), 142.73 (C=CH), 147.50 (C=N), 164.73 and 166.16 (2C=O, ester) ppm; MS: m/z (%) = 392 (M⁺, 4), 267 (4), 246 (4), 216 (20), 170 (76), 167 (100), 166 (80).

Dimethyl 2-(tert-butyliminomethylene)-3-(carbazol-9-yl)succinate (5a, C₂₃H₂₄N₂O₄)

Yellow oil, yield 0.29 g (37%), IR (KBr): $\bar{\nu} = 2035$ (C=C=N), 1741 and 1680 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.10$ (s, CMe₃), 3.57 and 3.98 (2s, 2OCH₃), 5.07 (s, CH), 7.27–8.12 (m, 8CH) ppm; ¹³C NMR: $\delta = 29.71$ (CMe₃), 51.82 and 53.02 (2OCH₃), 58.60 (CH), 62.11 (C=C=N), 62.18 (CMe₃), 109.96, 119.71, 120.29, and 125.90 (8CH), 123.35 and 139.75 (4C), 165.01 (C=C=N), 169.30, and 170.01 (2C=O, ester) ppm; MS: m/z (%) = 392 (M⁺, 4), 267 (4), 246 (4), 216 (20), 170 (76), 167 (100), 166 (80).

Di-tert-butyl 2-(tert-butylimino(carbazol-9-yl)methyl)but-2-enedioate (4b, C₂₉H₃₆N₂O₄)

White crystals (from diethyl ether), yield 0.85 g (90%), mp 240–241°C; IR (KBr): $\bar{\nu} = 1709$ and 1676 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.09$, 1.30 and 1.41 (3s, 3CMe₃), 6.22 (s, CH), 7.21–8.08 (m, 8CH) ppm; ¹³C NMR: $\delta = 27.76$, 28.37 and 29.64 (3CMe₃), 55.01 (NCMe₃), 80.30 and 82.58 (2OCMe₃), 110.58,

119.41, 120.31, and 125.81 (8CH), 123.35 and 139.51 (4C), 125.53 (CH vinyl), 140.02 (*C*=CH), 151.01 (C=N), 167.67 and 168.81 (2C=O, ester) ppm; MS: m/z (%) = 475 (M⁺ - 1, 2), 168 (20), 167 (100), 140 (30), 139 (42), 61 (56), 54 (52).

Di-tert-butyl 2-(tert-butyliminomethylene)-3-(carbazol-9-yl)succinate (5b, C₂₉H₃₆N₂O₄)

Yellow oil, yield 0.04 g (5%), IR (KBr): $\bar{\nu} = 2040$ (C=C=N), 1720 and 1678 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.01$, 1.25, and 1.34 (3s, 3CMe₃), 5.21 (s, CH), 7.21–8.08 (m, 8CH) ppm; ¹³C NMR: $\delta = 27.61$, 28.21, and 29.52 (3CMe₃); 54.88 (CH), 61.44 (C=C=N), 64.45 (NCMe₃), 74.02 and 85.02 (2OCMe₃), 110.38, 119.34, 119.97, and 125.53 (8CH), 123.03 and 139.21 (4C), 165.27 (C=C=N), 168.01, and 168.66 (2C=O, ester) ppm; MS: m/z (%) = 475 (M⁺ – 1, 2), 168 (20), 167 (100), 140 (30), 139 (42), 61 (56), 54 (52).

Dimethyl 2-(carbazol-9-yl(cyclohexylimino)methyl)but-2-enedioate (4c, C25H26N2O4)

White crystals (from diethyl ether), yield 0.48 g (58%), mp 142–144°C; IR (KBr): $\bar{\nu} = 1727$ and 1714 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.23-1.87$ (m, 5CH₂), 3.65 (m, CHN), 3.66 and 3.68 (2s, 2OCH₃), 6.44 (s, CH), 7.23–8.12 (m, 8CH) ppm; ¹³C NMR: $\delta = 23.43$, 25.72, 32.76, and 32.88 (5CH₂), 52.13 and 52.36 (20*CH*₃), 60.87 (CHN), 109.96, 120.69, 120.90, and 126.75 (8CH), 123.11 and 139.79 (4C), 123.64 (CH vinyl), 145.27 (*C*=CH), 145.81 (C=N), 164.64 and 165.93 (2C=O, ester) ppm; MS: m/z (%) = 417 (M⁺ – 1, 2), 238 (42), 220 (100), 167 (100), 166 (36), 113 (20), 52 (14).

Dimethyl 2-(cyclohexyliminomethylene)-3-(carbazol-9-yl)succinate (**5c**, C₂₅H₂₆N₂O₄)

Yellow oil, yield 0.31 g (37%), IR (KBr): $\bar{\nu} = 2045$ (C=C=N), 1731 and 1683 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.22-1.67$ (m, 5CH₂), 3.34 (m, CHN), 3.57 and 3.97 (2s, 2OCH₃), 5.32 (s, CH), 7.20–8.10 (m, 8CH) ppm; ¹³C NMR: $\delta = 23.88$, 25.44, 31.93, and 33.22 (5CH₂), 51.74 and 52.63 (2OCH₃), 54.21 (CH), 60.64 (C=C=N), 60.73 (CHN), 109.96, 119.67, 120.11, and 125.84 (8CH), 123.11 and 139.59 (4C), 165.35 (C=C=N), 169.34 and 169.96 (2C=O, ester) ppm; MS: m/z (%) = 417 (M⁺ – 1, 2), 238 (42), 220 (100), 167 (100), 166 (36), 113 (20), 52 (14).

Di-tert-butyl 2-(carbazol-9-yl(cyclohexylimino)methyl)but-2-enedioate (4d, $C_{31}H_{38}N_2O_4$)

White crystals (from diethyl ether), yield 0.29 g (58%), mp 130–132°C; IR (KBr): $\bar{\nu} = 1708$ and 1687 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.12-1.66$ (m, 5CH₂), 1.30 (s, 2CMe₃), 3.60 (m, CHN), 6.25 (s, CH), 7.14–8.06 (m, 8CH); ¹³C NMR: $\delta = 23.76$, 25.23, 27.94, 30.87, and 32.78 (5CH₂), 27.79 and 28.38 (2CMe₃), 60.74 (CHN), 80.84 and 82.63 (2OCMe₃), 110.41, 119.36, 120.63, and 125.51 (8CH), 123.49 and 139.75 (4C), 124.99 (CH vinyl), 139.43 (C=CH), 146.43 (C=N), 167.77 and 168.71 (2C=O, ester); MS: m/z (%) = 502 (M⁺, 2), 341 (8), 280 (12), 217 (18), 216 (36), 167 (38), 166 (100), 142 (30), 82 (58), 54 (46).

Di-tert-butyl 2-(cyclohexyliminomethylene)-3-(carbazol-9-yl)succinate (5d, C₃₁H₃₈N₂O₄)

Yellow oil, yield 0.18 g (37%), IR (KBr): $\bar{\nu} = 2040$ (C=C=N), 1719 and 1642 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.22-1.66$ (m, 5CH₂), 1.40 (s, 2CMe₃), 3.41 (m, CHN), 5.44 (s, CH), 7.14–8.06 (m, 8CH) pp; ¹³C NMR: $\delta = 23.82$, 25.31, 28.10, 31.61, and 32.92 (5CH₂), 27.56 and 28.09 (2CMe₃), 55.30 (CH), 60.06 (C=C=N), 62.79 (CHN), 75.94 and 83.32 (2OCMe₃), 110.12, 119.34, 120.49, and 126.15 (8CH), 123.01 and 139.42 (4C), 164.34 (C=C=N), 168.45 and 168.75 (2C=O, ester) ppm; MS: m/z (%) = 502 (M⁺, 2), 341 (8), 280 (12), 217 (18), 216 (36), 167 (38), 166 (100), 142 (30), 82 (58), 54 (46).

Synthesis of Highly Functionalized 1-Azadienes and Ketenimines

Dimethyl 2-(tert-butylimino(indol-1-yl)methyl)but-2-enedioate (4e, C₁₉H₂₂N₂O₄)

White crystals (from diethyl ether), yield 0.39 g (58%), mp 91–92°C; IR (KBr): $\bar{\nu} = 1717$ and 1657 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.36$ (s, CMe₃), 3.60 and 3.79 (2s, 2OCH₃), 6.51–8.63 (m, 6CH_{arom}), 7.21 (s, CH vinyl) ppm; ¹³C NMR: $\delta = 30.95$ (CMe₃), 52.53 and 53.39 (2OCH₃), 55.41 (CMe₃), 105.07, 116.34, 120.44, 121.83, 123.42, and 125.34 (6CH), 130.41 and 135.86 (2C), 132.25 (CH vinyl), 140.22 (C=CH), 142.14 (C=N), 163.86 and 164.48 (2C=O, ester) ppm; MS: m/z (%) = 343 (M⁺ + 1, 2), 302 (8), 226 (28), 213 (26), 170 (90), 169 (8), 138 (26), 118 (32), 117 (100), 54 (100).

Dimethyl 2-(tert-butyliminomethylene)-3-(indol-1-yl)succinate (5e, C₁₉H₂₂N₂O₄)

Yellow oil, yield 0.25 g (37%), IR (KBr): $\bar{\nu} = 2040$ (C=C=N), 1729 and 1687 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.32$ (s, CMe₃), 3.69 and 3.73 (2s, 2OCH₃), 6.05 (s, CH), 6.52–7.59 (m, 6CH) ppm; ¹³C NMR: $\delta = 30.06$ (CMe₃), 51.89 and 52.86 (2OCH₃), 55.33 (CH), 62.51 (C=C=N), 63.19 (CMe₃), 102.73, 109.62, 119.97, 120.99, 121.98, and 126.10 (6CH), 128.79 and 136.05 (2C), 164.85 (C=C=N), 169.45 and 169.45 (2C=O, ester) ppm; MS: m/z (%) = 343 (M⁺ + 1, 2), 302 (8), 226 (28), 213 (26), 170 (90), 169 (8), 138 (26), 118 (32), 117 (100), 54 (100).

Dimethyl 2-(tert-butylimino(pyrrol-1-yl)methyl)but-2-enedioate (4f, C₁₅H₂₀N₂O₄)

White crystals (from diethyl ether), yield 0.26 g (90%), mp 95–97°C; IR (KBr): $\bar{\nu} = 1661$ and 1617 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.18$ (s, CMe₃), 3.61 and 3.75 (2s, 2OCH₃), 7.11 (s, CH), 6.11–7.07 (m, 4CH) ppm; ¹³C NMR: $\delta = 30.59$ (CMe₃), 52.36 and 53.28 (2OCH₃), 58.34 (CMe₃), 110.26 and 118.27 (4CH), 132.44 (CH vinyl), 140.03 (C=CH), 140.36 (C=N), 163.63 and 164.20 (2C=O, ester) ppm; MS: m/z (%) = 292 (M⁺, 2), 216 (6), 80 (4), 54 (40), 39 (38), 35 (100).

Di-tert-butyl 2-(tert-butylimino(pyrrol-1-yl)methyl)but-2-enedioate (4g, C₂₁H₃₂N₂O₄)

White crystals (from diethyl ether), yield 0.67 g (90%), mp 138–140°C; IR (KBr): $\bar{\nu} = 1661$ and 1617 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.26$, 1.27, and 1.40 (3s, 3CMe₃), 6.93 (s, CH), 6.13–7.12 (m, 4CH) ppm; ¹³C NMR: $\delta = 27.57$, 27.71, and 30.87 (3CMe₃), 54.76 (NCMe₃), 82.94 and 83.31 (2OCMe₃), 109.94 and 118.32 (4CH), 133.60 (CH vinyl), 139.95 (C=CH), 140.05 (C=N), 162.84 and 162.92 (2C=O, ester) ppm; MS: m/z (%) = 367 (M⁺, 2), 311 (6), 198 (22), 143 (40), 142 (100), 54 (100).

Dimethyl 2-(cyclohexylimino(pyrrol-1-yl)methyl)but-2-enedioate (4h, C₁₇H₂₂N₂O₄)

White crystals (from diethyl ether), yield 0.27 g (85%), mp 86–88°C; IR (KBr): $\bar{\nu} = 1720$ and 1657 (C=O) cm⁻¹; ¹H NMR: $\delta = 1.25-1.85$ (m, 5CH₂), 3.20 (m, CHN), 3.67 and 3.81 (2s, 2OCH₃), 7.25 (s, CH), 6.22–7.21 (m, 4CH) ppm; ¹³C NMR: $\delta = 23.98$, 25.45, 33.10, and 33.34 (5CH₂), 52.12 and 53.05 (2OCH₃), 59.70 (CHN), 110.43 and 118.21 (4CH), 132.46 (CH vinyl), 136.86 (*C*=CH), 143.66 (C=N), 163.25 and 163.41 (2C=O, ester) ppm; MS: m/z (%) = 319 (M⁺ + 1, 2), 253 (8), 170 (100), 169 (8), 138 (32), 82 (36), 52 (46).

References

- [1] Ugi I (1982) Angew Chem Int Ed Engl 21: 810
- [2] Ugi I, Domling A, Horl W (1994) Endeavour 18: 115
- [3] Ugi I, Lohberger S, Karl R (1991) In: Trost BM, Fleming I (eds) Comprehensive Organic Synthesis, vol 2. Pergamon Oxford, p 1033
- [4] Domling A, Ugi I (2000) Angew Chem Int Ed Engl 39: 3168

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- [5] Suzuki Y, Obata N, Takizawa T (1970) Tetrahedron Lett 11: 2667
- [6] Suzuki Y, Litaka Y (1971) Bull Chem Soc Jpn 44: 56
- [7] Oakes TR, Donovan TJ (1973) J Org Chem 38: 1319
- [8] Dillinger HJ, Fengler G, Schumann D, Winterfeldt E (1974) Tetrahedron 30: 2553
- [9] Junjappa H, Saxena MK, Ramaiah D, Loharay BB, Rath NP, George MV (1998) J Org Chem 63: 9801
- [10] Nair V, Vinod AU (2000) J Chem Soc Chem Commun 1019
- [11] Nair V, Vinod AU, Rajesh C (2001) J Org Chem 66: 4427
- [12] Beaudegnies R, Ghosez L (1994) Tetrahedron Asymmetry 5: 557
- [13] Gilchrist TL, Gonsalves AMdAR, Pinho e Melo TMVD (1996) Pure Appl Chem 68: 859
- [14] Walborsky HM, Persiasamy MP (1983) In: Patai S, Rappoport Z (eds) The Chemistry of Functional Groups, Suppl C, chapt 20. Wiley, New York, p 835
- [15] Marcaccini S, Torroba T (1993) Org Prep Proced Intl 25: 141
- [16] Moderhack D (1985) Synthesis 1083

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